1. (6 pts each) Each of the following reaction schemes contains some sort of error. In each case, clearly identify all the error(s) and correct it/them. The errors may be in any element of the transformation (starting materials, reagents, products).

2. (6 pts each) Predict the major product of the following reactions. Mechanisms and explanations are not necessary.

a. 
$$CH_3COOH$$
 $CH_3COOH$ 
 $CH_3COOH$ 
 $CH_3$ 
 $CH_3$ 

c. 
$$H_3O^+$$
 $PH 5$ 

3. (12 pts) Outline two methods for preparing isopropyl methyl ether. Which method is likely to give a better yield of the ether? Explain.

\* better yield less Steric hindrance for SN2 when alkyl halide is attack

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4. (14 pts) Only one of the products to the right can be formed by an aldol reaction. Show the necessary reactant(s) and write a complete mechanism for the reaction you propose.

ONLY

- 5. (10 pts) What reaction or series of reactions would you use to carry out the following transformation. Mechanisms are not necessary.

6. (12 pts) Many of the reactions that were covered in chapters 16 and 17 (reactions of aldehydes and ketones) could be catalyzed by both acid and base. Give an example of one reaction that could only be catalyzed by either acid or base (not both). Clearly indicate which catalyst works, and explain why the other catalyst does not work. A full mechanism is not necessary, but your answer must include relevant structures in addition to your explanation.

Several options:

- 1. cyanohydrin-only base to make aN nucleophile
- 2. hemiacetal to acetal-only acid 3 of converted to a better LG
- 4. making kinetic enolate only base use LDA as strong bulky base

7. (16 pts) Show how you would synthesize the following compounds starting with saturated alcohols containing four or fewer carbons, oxirane, plus any inorganic reagents needed. You may also use organic compounds (eg TsCl, PPh3, etc) as long as their carbons are not incorporated into the target. A molecule made in one synthesis can be used in another synthesis without "resynthesizing" it.