1. (6 pts each) Predict the major product of the following reactions. Mechanisms and explanations are not necessary.

a. 
$$H_{2O}$$
  $H_{2O}$   $+ CO_{2}$ 

b. 
$$\frac{\text{excess CH}_{3}I}{\text{NH}_{2}} \xrightarrow{\text{1. Ag}_{2}O, \text{ H}_{2}O} + \text{NCH}_{3})_{3}$$

2. (14 pts) Only one of the keto esters to the right can be formed by a Claisen (Dieckmann) condensation. Determine which one, show the necessary reactant(s), and write a complete mechanism for the reaction you propose. For partial credit, you may show any correct Claisen mechanism.

3. (10 pts) When 1,2-dichloro-3-cyanobenzene is treated obtained. Write an equation for the substitution, showing the product you expect. Give a mechanism to account for the formation of your proposed product.

4. (10 pts each) Show how you would convert nitrobenzene into the following compounds. Assume any reagents you need are available. Mechanisms and explanations are not required. Answer any TWO of the three parts.

Noz 
$$OCH_3$$

(a)  $OCH_3$ 

(b)  $OCH_3$ 
 $OCH_3$ 

5. (10 pts) The amide shown to the right exhibits an IR Relative to the normal stretching range for an amide carbonyl (1680-1700 cm<sup>-1</sup>), this value is quite high. Using a combination of drawings and words amide exhibits an unusually high stretching frequency.

3



6. a. (8 pts) Which of the phenols to the right is the stronger acid. Justify your answer using both words and drawings. Show any resonance structures that are central to your argument.

B conjugate base has extra resonance W/NO2

Since conj base is more stable,

B is the stronger acid

b. (8 pts) Which of the amines to the right is the stronger base? Justify your answer using both words and drawings. Show any resonance structures that are central to your argument.

B lacks the base weakening resonance effect of the avantic ving and the base weakening e-withdrawing effect of the aronatic vix.

(12 pts) Provide a mechanism for the reaction shown below.