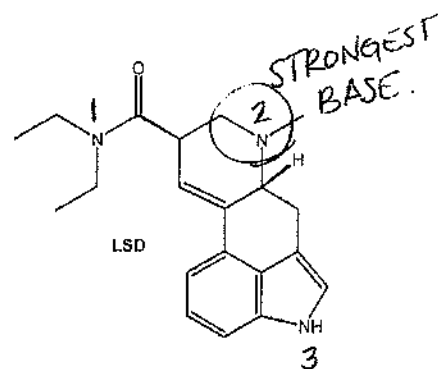



1. (14 pts) (a) Identify the most basic nitrogen in LSD. (b) Clearly explain why that nitrogen is the most basic and why the other two nitrogens are less basic. You must draw representative (not all!) resonance structures that are central to your argument.

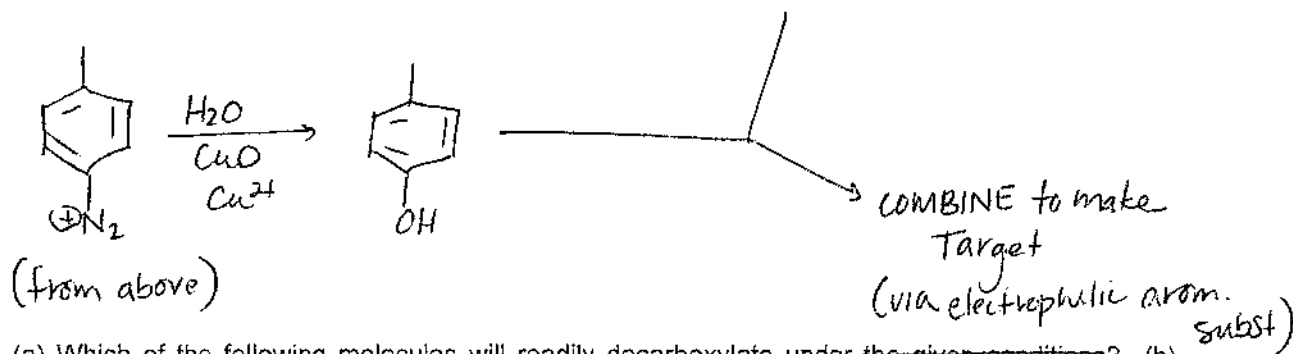
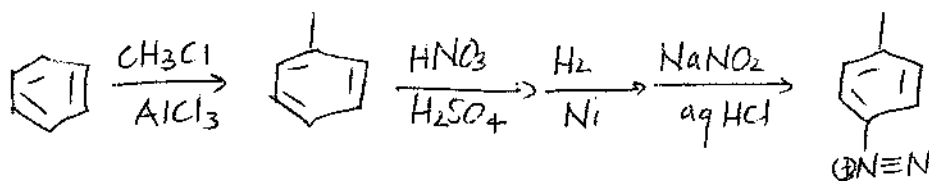
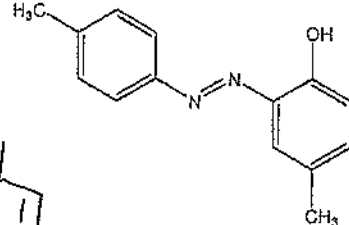


N #1. Lone pair is tied up by inductive effect of $C=O$
 $\ddot{N} \rightarrow C$ and resonance with $C=O$ ($\overset{+}{N}=\overset{-}{C}$)

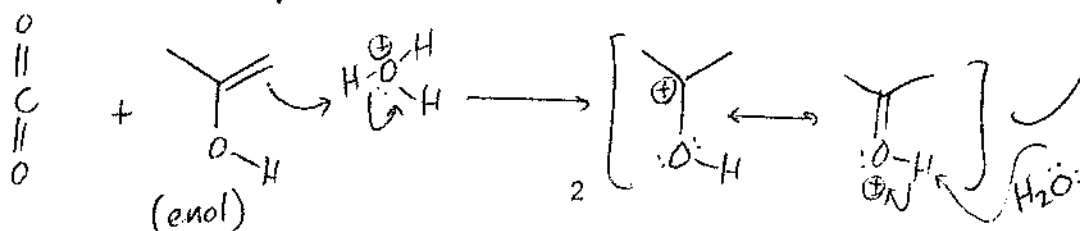
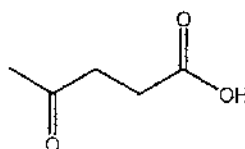
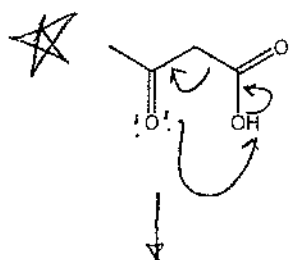
N#2. 3R groups are e⁻ donating inductively $\nearrow \ddot{N} \leftarrow$ making N a stronger base. No resonance to tie up lone pair \uparrow

N#3. sp^2 groups are e^- -withdrawing so lone pair is less available
Resonance ties up lone pair. eg.  \leftrightarrow others

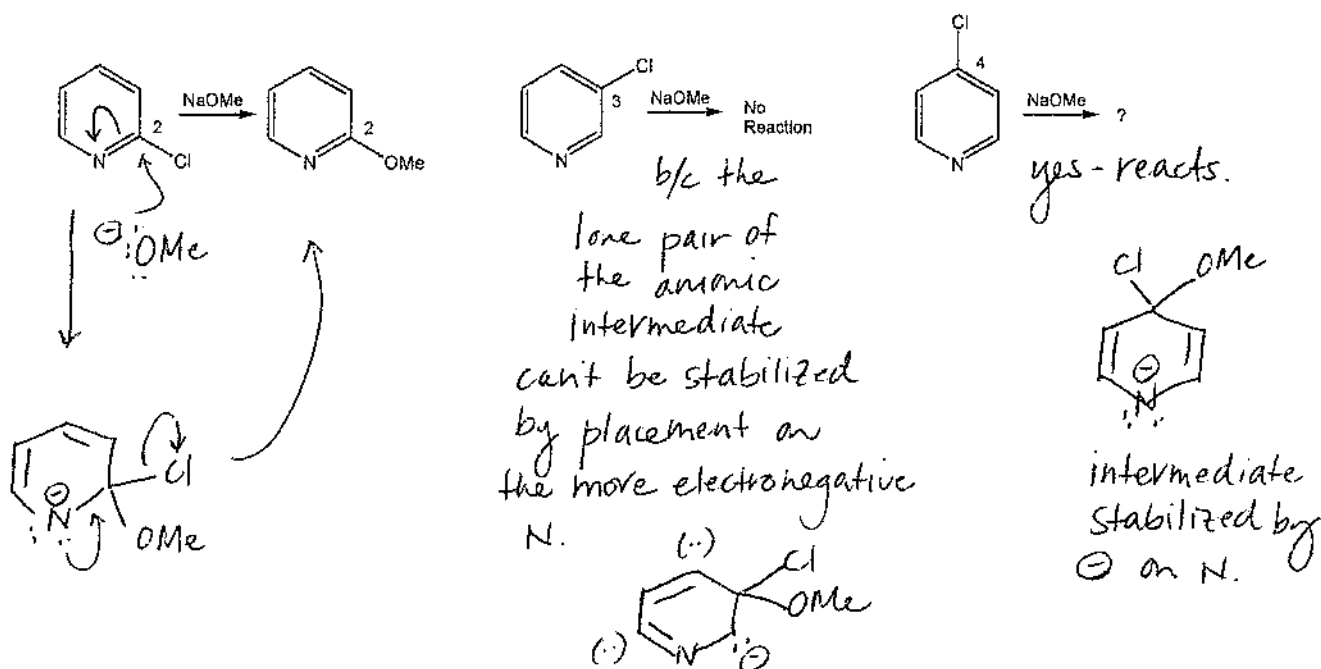
2. (12 pts) Outline a synthesis of the following compound. The only aromatic compound you are allowed to use as a reactant is benzene. You can assume that any other nonaromatic reagents are available. Mechanisms are not required.



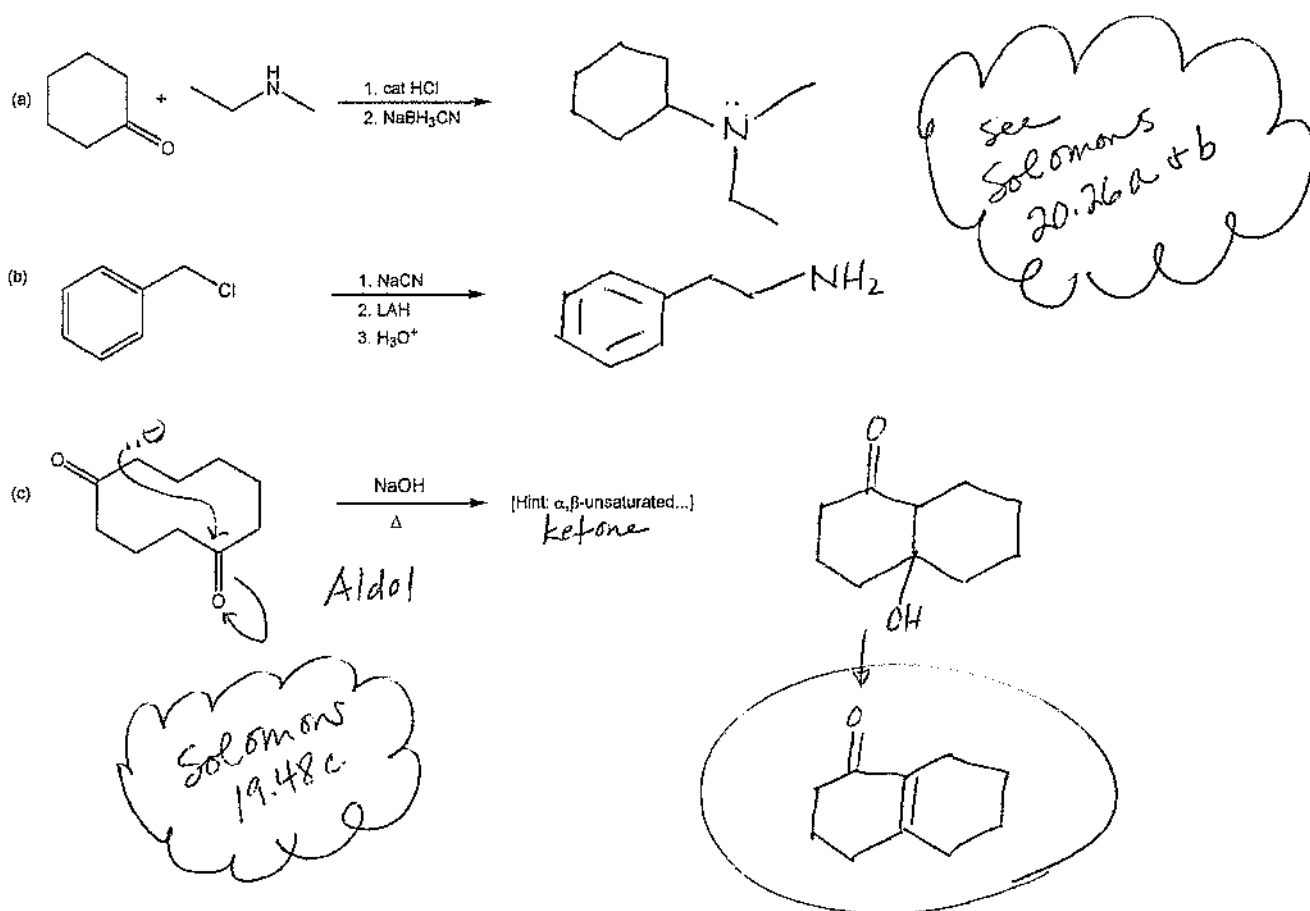
3. (12 pts) (a) Which of the following molecules will readily decarboxylate under the given conditions? (b) Draw a mechanism to illustrate the decarboxylation and formation of the major product obtained from the successful reaction. Use H_2O , H_3O^+ as needed.



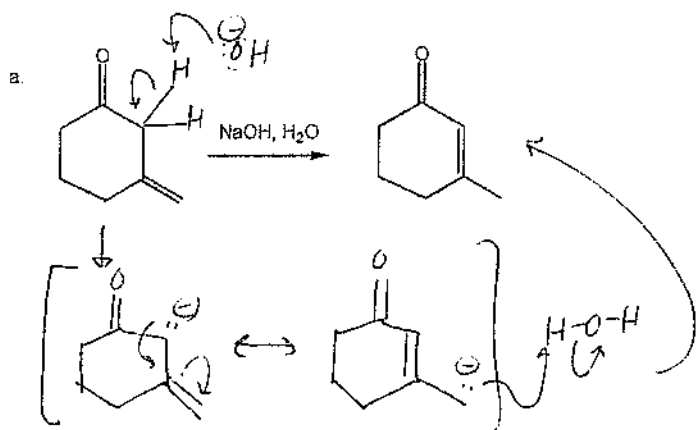
4. (12 pts) 2-chloropyridine reacts with sodium methoxide via an addition-elimination mechanism. (a) Draw a mechanism to show why 2-chloropyridine successfully reacts. (b) Explain why 3-chloropyridine does not react. (b) Predict whether 4-chloropyridine will successfully react with sodium methoxide. Explain.



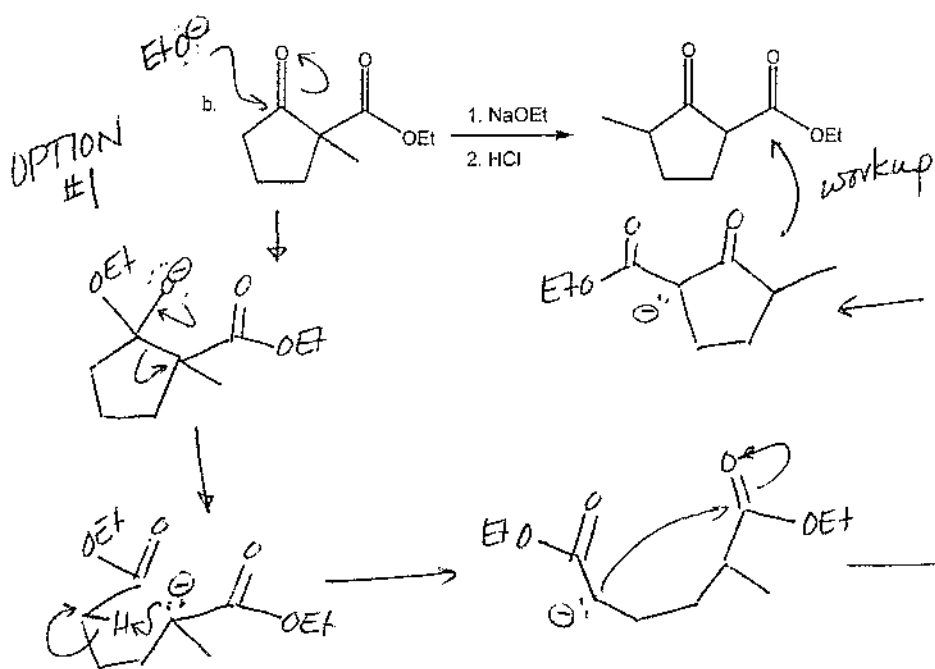
5. (16 pts) Provide the major organic product from each of the following reactions. No mechanisms required.



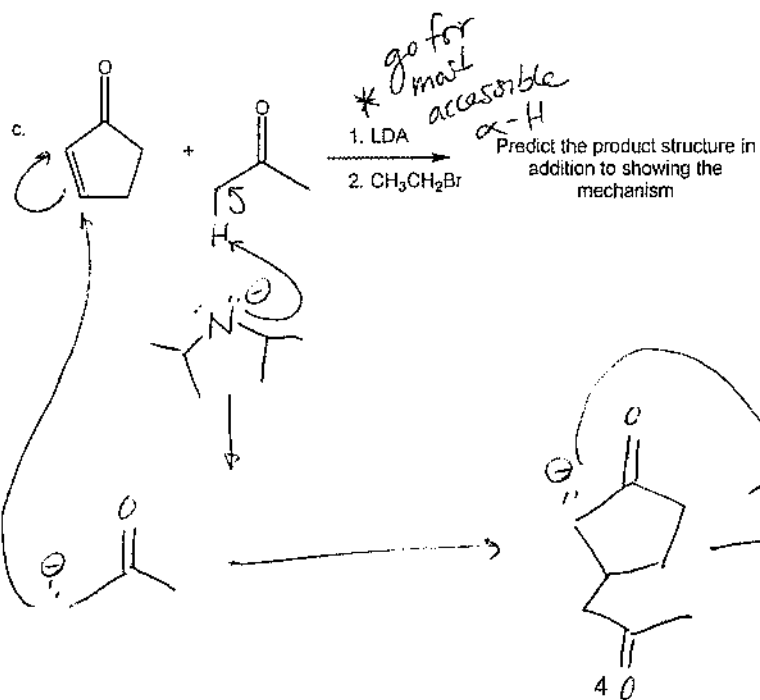
6. (34 pts) Write a full curved arrow mechanism for each of the following reactions:



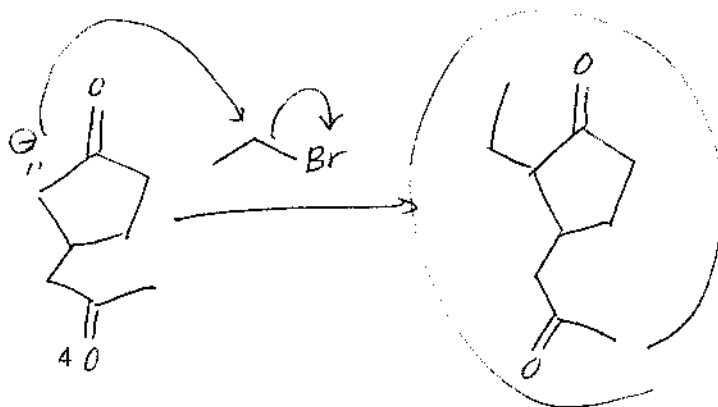
Solomons
18.20a



Solomons
19.31



* SEE LAST PAGE FOR 2 OTHER CORRECT MECHANISMS



OPTION
#2

